

New Honeycomb Iridium (V) Oxides: NaIrO_3 and $\text{Sr}_3\text{CaIr}_2\text{O}_9$.

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Abstract

We report the structures and physical properties of two new iridates, NaIrO_3 and $\text{Sr}_3\text{CaIr}_2\text{O}_9$, both of which contain continuous two-dimensional honeycomb connectivity. NaIrO_3 is produced by room temperature oxidative deintercalation of sodium from Na_2IrO_3 , and contains edge-sharing IrO_6 octahedra that form a planar honeycomb lattice. $\text{Sr}_3\text{CaIr}_2\text{O}_9$, produced via conventional solid-state synthesis, hosts a buckled honeycomb lattice with novel corner-sharing connectivity between IrO_6 octahedra. Both of these new compounds are comprised of Ir^{5+} ($5d^4$) and exhibit negligible magnetic susceptibility. They are thus platforms to investigate the origin of the nonmagnetic behavior exhibited by Ir^{5+} oxides, and provide the first examples of a $J = 0$ state on a honeycomb lattice.

1 Introduction

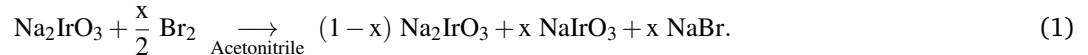
Research into 5d transition metal compounds has blossomed in recent years due to the potential for new electronic and magnetic phenomena to emerge from the interplay between crystal field effects, electronic correlations (Hubbard U) and strong spin-orbit coupling (SOC). Iridium is an appealing element in which to study these physics due to its tendency to adopt a 4+ oxidation state, where the ground electronic configuration $5d^5$ is predicted to yield a $J = \frac{1}{2}$ electronic state in the presence of strong SOC¹. Further, iridates are known to crystallize in numerous structure types, each with unique Ir-Ir connectivity, and this fact has made them an active area of study²⁻⁵. In aggregate, recent literature on the magnetic and electronic properties of iridates are unified by a desire to probe the influence of strong SOC and U on electronic behavior⁶. Iridium-based honeycombs, in particular the $A_2\text{IrO}_3$ ($A = \text{Li/Na}$) family, have fallen under intense scrutiny following predictions that they might host a spin-liquid ground state⁷⁻¹⁴. Efforts to produce new iridium oxide honeycomb lattices have intensified¹⁵⁻¹⁷, but 4d and 5d materials containing continuous honeycomb connectivity remain scarce.

Here we report the synthesis and characterization of two new honeycomb iridates based on Ir^{5+} ($5d^4$). The first, NaIrO_3 is produced via *chimie douce* oxidative deintercalation of sodium from Na_2IrO_3 , which preserves the planar honeycomb network of edge-sharing IrO_6 octahedra. The second, $\text{Sr}_3\text{CaIr}_2\text{O}_9$, is a 2:1 ordered perovskite, which shares its structure with $\text{Sr}_3\text{CaRu}_2\text{O}_9$ ¹⁸, and consists of layers of IrO_6 octahedra forming a buckled honeycomb lattice. Magnetic susceptibility data collected on both compounds demonstrate weak temperature-independent magnetism, and resistivity measurements show semiconducting behavior consistent with hopping conductivity. These compounds are thus ideal to investigate the interplay between non-cubic crystal fields and SOC. By comparison with related 4d and 5d honeycomb compounds with partially filled t_{2g} subshells, we find that d electron count correlates strongly with the magnetic behavior, and present a thorough structural analysis of $\text{Sr}_3\text{CaIr}_2\text{O}_9$ and $\text{Sr}_3\text{CaRu}_2\text{O}_9$ that ultimately suggests the correlation is driven by SOC.

2 Results

2.1 Syntheses and Structures

NaIrO_3 was prepared via the oxidative deintercalation reaction:



The 1:1 relationship between oxidant consumption and Na^+ deintercalation of equation 1 was confirmed by laboratory X-ray powder diffraction (XRPD) data collected on powders after vacuum evaporation of the solvent, which showed crystalline NaBr as the only side product. In contrast to the precursor material Na_2IrO_3 which degrades rapidly in laboratory air¹⁹, NaIrO_3 is stable under ambient conditions, as XRPD data showed no evidence of decomposition after several months of air exposure. However, NaIrO_3 degrades rapidly upon heating above 200°C in air, as evidenced by the broadening of

diffraction peaks and appearance of new reflections. Attempts at producing intermediate stoichiometries (i.e. $\text{Na}_{2-x}\text{IrO}_3$) were unsuccessful, and instead produced two-phase samples containing a mixture of Na_2IrO_3 and NaIrO_3 . Further, no phase width was observed in NaIrO_3 , as reactions performed with excess bromine produced only NaIrO_3 with no observable difference in lattice parameters or physical properties.

The structure of NaIrO_3 was solved in space group $P\bar{1}$ via Rietveld refinement to laboratory XRPD data (Fig. 1(a)). Final structural parameters are available in the SI. Solution of the structure in higher symmetry cells was prevented by the high degree of inter-plane stacking disorder, which arises due to the fact that there are several possible translations that can take place in the honeycomb plane from one layer to another, each of which generates a different stacking pattern at minimal energetic cost. Stacking disorder is a well known structural perturbation in many layered compounds, and evidence of this disorder can be plainly seen in the poorly-fit peak broadening and incorrect intensities present at low diffraction angles. Further evidence of this structural disorder is provided by synchrotron XRPD data (Fig. 1 (a, inset)), which show highly asymmetric $\langle 0\ 0\ 1 \rangle$ peaks—this may occur as a consequence of decomposition in the intense synchrotron beam. Using DIFFaX²⁰, we modeled the diffraction patterns for three distinct stacking variants. When viewed from perpendicular to the honeycomb plane, these stacking variants can be described based on which atoms in the honeycomb plane eclipse (sit directly above) the Na ion in the adjacent plane. The three primary stacking possibilities are: 1) Na-Ir eclipsed; 2) Na-Na fully eclipsed, which produces infinite Na channels; 3) staggered, in which none of the atoms in the honeycomb plane eclipse those in the adjacent plane, which is the pattern adopted by the parent material Na_2IrO_3 . The simulated diffraction patterns for these stacking arrangements, along with XRPD data collected on a sample of Na_2IrO_3 are shown in figure Fig. 1(b). The observed XRPD pattern closely resembles the simulation obtained for the Na-Ir eclipsed variant, but with noticeable broadening and attenuation of several peaks. A fourth pattern was simulated based on a model consisting of 95% Na-Ir eclipsed stacking with a 5% probability of a fully-eclipsed stacking fault. This model produces a good qualitative agreement with the raw XRPD data, suggesting that the fully eclipsed stacking variant may only be slightly less energetically favorable than the Na-Ir eclipsed variant. The presence of the fully eclipsed stacking fault suggests that further synthetic work may yield control over stacking order in this compound. For an in-depth discussion on stacking disorder in honeycomb iridates, see ref.²¹. This is the second known structure with the formula NaIrO_3 —a post-perovskite structure with the same formula is obtained when synthesized under high-pressure and temperature²². Neither structural polymorph of NaIrO_3 can be synthesized using conventional solid-state techniques, a fact that underscores the novelty and relative instability of the Ir^{5+} oxidation state.

The structure of $\text{Sr}_3\text{CaIr}_2\text{O}_9$ was solved in space group $P2_1/c$ (14) via Rietveld refinement to neutron powder diffraction (NPD) and XRPD data (Fig. 2 (a)), and was found to be isostructural to $\text{Sr}_3\text{CaRu}_2\text{O}_9$ ¹⁸. Final structural parameters are available in the SI. The monoclinic structure of $\text{Sr}_3\text{CaIr}_2\text{O}_9$, visible in Fig. 2 (b), can be described as a 2:1 ordered perovskite with the formula $\text{A}_3\text{BB}'_2\text{O}_9$, where IrO_6 and CaO_6 octahedra share corners to form the perovskite lattice and Sr^{2+} cations occupy the 12-fold coordinate A sites. In contrast to the $\text{Ba}_3\text{M}\text{Ir}_2\text{O}_9$ ($\text{M} = \text{Mg}, \text{Ca}, \text{Sc}, \text{Ti}, \text{Zn}, \text{Sr}, \text{Zr}, \text{Cd}, \text{In}, \text{etc.}$) family of 6H-perovskite-like structures, which host face-sharing Ir_2O_9 dimers²³, the IrO_6 octahedra of $\text{Sr}_3\text{CaIr}_2\text{O}_9$ share

corners to form a buckled honeycomb lattice in the b–c plane. While the structure of $\text{Sr}_3\text{CaIr}_2\text{O}_9$ may seem odd given the apparent wealth of other Barium-based compounds with analogous stoichiometry, it is actually unsurprising that a conventional perovskite lattice is formed in this case due to the improved match in ionic radii between Sr and Ir. The same is true of $\text{Sr}_3\text{CaRu}_2\text{O}_9$, which also has a similar collection of Ba-based 6H-perovskite-type cousins. While many different perovskites containing iridium have been reported in addition to the widely studied Sr_2IrO_4 , this seems to be a rare example of an iridium-based perovskite containing honeycomb connectivity. Furthermore, the corner-sharing bonding motif of the honeycomb lattice is the first of its kind in iridates, and thus opens a new avenue along which to search for new honeycomb materials that are free from stacking disorder and are air-stable.

The corefinement of the $\text{Sr}_3\text{CaIr}_2\text{O}_9$ structure to both XRPD and NPD data yielded stable oxygen positions that can be used to investigate the honeycomb connectivity in detail (Fig. 2 (b)). As a result of octahedral tilting, the Ir–O–Ir bond angles between nearest-neighbor sites on the honeycomb lattice are significantly less than the ideal 180° expected for an undistorted perovskite. The Ir–O bond lengths are also distorted: for the first iridium site, three adjacent Ir–O bonds are significantly shorter than the overall average of 1.99 \AA , while the opposite three are significantly longer. A similar “three short, three long” pattern is also observed to a lesser degree on the second iridium site, where the three short bonds form a plane that is roughly perpendicular to the plane formed by the longer three bonds. This pattern results in significant variations in the bridging Ir–O–Ir bond lengths: the shortest Ir–Ir distance ($3.96(3) \text{ \AA}$) is formed via Ir–O bonds of very similar lengths ($2.04(4)$ & $2.05(3) \text{ \AA}$, respectively), while the longest ($d_{\text{Ir–Ir}} = 4.01(1) \text{ \AA}$) is formed via Ir–O bonds of very disproportionate lengths ($2.14(3)$ & $1.92(4) \text{ \AA}$, respectively). These non-uniform exchange interactions result in an effective dimerization between Ir sites even though the distances between adjacent iridium sites on the honeycomb lattice are remarkably similar. These observations are consistent with a non-spherical perturbation to the d-electron states of Ir $5+$, which will be discussed in detail in the next section.

2.2 Physical Properties

Fig. 3 shows magnetic susceptibility data collected on NaIrO_3 and $\text{Sr}_3\text{CaIr}_2\text{O}_9$, along with linear fits to the χ_0 -corrected inverse susceptibilities (inset). In contrast to Na_2IrO_3 , which has a magnetic moment of $2.0(1) \mu_B$ per Ir site and a Weiss temperature $\theta_W = -159(3) \text{ K}$, honeycomb NaIrO_3 shows only a small temperature-independent susceptibility $\chi_0 = 3.29 \times 10^{-4} \text{ emu/mol Ir}$ in the high temperature regime and a Curie tail at low temperatures. A Curie-Weiss fit to the magnetic susceptibility data (Fig. 3 (inset)) yields a Curie constant of $C = 4.3(1) \times 10^{-3} \text{ emu/mol Ir K}$, and a Weiss temperature $\theta_W = 0(2) \text{ K}$. Similar magnetic behavior is observed in $\text{Sr}_3\text{CaIr}_2\text{O}_9$ (Fig. 3). $\text{Sr}_3\text{CaIr}_2\text{O}_9$ was initially obtained as a phase-pure powder after 6 short heatings, and a Curie-Weiss fit to the magnetic susceptibility data collected on this sample (Fig. 3 (inset)) yielded a temperature-independent susceptibility of $\chi_0 = 2.6(1) \times 10^{-3} \text{ emu/mol Ir}$, a Curie constant $C = 9.3(1) \times 10^{-3} \text{ emu/mol Ir K}$, and a Weiss temperature $\theta_W = -8(5) \text{ K}$. Optimization of the stoichiometry and heating schedule for $\text{Sr}_3\text{CaIr}_2\text{O}_9$ (See E.S.I.) yielded a sample whose diffraction peaks were significantly sharper and more intense, consistent with an improvement in crystallinity and homogeneity. This improvement resulted in a substantial

reduction of C, Θ_W and χ_0 ($4.6(1) \times 10^{-3}$ emu/mol Ir K, -8(5) K, and 3.29×10^{-4} emu/mol Ir, respectively). In all samples, the small observed magnetic moments are likely due to dilute magnetic impurities or orphan spins—roughly 1 % of free $S = \frac{1}{2}$ spins in the bulk could account for the observed susceptibility. Furthermore, the small Weiss temperatures observed from all datasets indicate that the magnetic electrons are non-interacting, which lends further weight to the argument for magnetic defects and impurities rather than intrinsic magnetism. These results are thus consistent with a $J = 0$ state in the Ir^{5+} metal centers of both compounds. Furthermore, the magnetic moments observed for Ir^{5+} in these compounds are consistent with what is observed in many other Ir^{5+} compounds (Table 1).

The observed temperature-independent susceptibility is not due to delocalization of charge carriers to form a metal, as is the case in other iridates such as $[\text{Rb/K}]\text{Ir}_4\text{O}_8$ ^{24,25}. Fig. 4 shows electrical resistivity data collected on polycrystalline bar of NaIrO_3 and $\text{Sr}_3\text{CaIr}_2\text{O}_9$. In both datasets, sample resistance increases on cooling, diverging rapidly below $T \sim 100$ K, and exceeding the detection limit of the instrument below $T \sim 75$ K in the case of NaIrO_3 . A simple Arrhenius-like activation barrier did not yield a good fit to the resistivity data. Instead, the behavior of both NaIrO_3 and $\text{Sr}_3\text{CaIr}_2\text{O}_9$ is well described by a variable-range hopping model. The conductivity is consistent with hopping in two or three dimensions: a plot of $\ln \rho$ vs. $T^{-\frac{1}{3}}$ (Fig. 4(inset)) or $\ln \rho$ vs. $T^{-\frac{1}{4}}$ (not shown) yields a linear relationship. This is similar to what is found for Na_2IrO_3 and indeed many other 4d and 5d honeycombs, and consistent with the materials structure^{3,37}.

3 Discussion

Both NaIrO_3 and $\text{Sr}_3\text{CaIr}_2\text{O}_9$ are important new additions to a select few 4d and 5d compounds that exhibit honeycomb connectivity between MO_6 octahedra, tabulated in Table 2. For a given d electron count, there is little difference in observed magnetic moments between the 4d and 5d groups, despite the fact that both crystal field and SOC energy scales change significantly when moving down the periodic table. While the magnetic moments observed for the d^5 configurations are unsurprising given the expected magnetic moment of $\mu \sim 1.9\mu_B$ for a $S = \frac{1}{2}$ system, the weak temperature-independent magnetism observed for both $4d^4 \text{Ru}^{4+}$ and $5d^4 \text{Ir}^{5+}$ is an unexpected result in the low-spin octahedral crystal field case, which should have 2 unpaired electrons and total spin $S = 1$. Fig. 5 shows three possible origins of a nonmagnetic state in an Ir^{5+}O_6 (Ru^{4+}O_6) octahedron. One possibility is that direct overlap between adjacent Ir sites generates new molecular orbitals, which removes the threefold degeneracy of the t_{2g} manifold and results in a $S = 0$ state, a scenario that has been proposed for Li_2RuO_3 based on the significant variation in Ru-Ru bondlengths²⁶. The second possibility is that distortions from perfect octahedral symmetry remove the degeneracy of the t_{2g} manifold locally (i.e. a Jahn-Teller distortion), thus resulting in a completely filled, twofold degenerate ground state. Such distortions are thought to be highly important in the Ir^{5+} perovskites $[\text{Sr/Ba}]_2\text{IrO}_4$, and may also be driven by extended crystal field effects (i.e. interactions beyond nearest-neighbor oxygen and iridium atoms)^{27,28}. The final possibility outlined here is that SOC is the strongest perturbation to the cubic crystal field in Ir^{5+} , which is consistent with electronic structure calculations performed by Phelan *et al.* on $\text{Sr}_x\text{La}_{11-x}\text{Ir}_4\text{O}_{24}$ ⁵. While the computational route invariably leads to the conclusion that SOC produces a nonmagnetic state in Ir^{5+} , one can also reach this conclusion on paper using group theory. To understand how SOC pro-

duces a $J = 0$ state in this case, one must generate appropriate term symbols for electronic states in the presence of SOC by referencing the double group for O_h symmetry. In addition to the five irreducible representations of standard character table for group O_h , the double group has six new irreducible representations ($\Gamma_6^{+/-}$, $\Gamma_7^{+/-}$, and $\Gamma_8^{+/-}$)²⁹. By referencing to the O_h double group, one finds that SOC splits the Γ_{25}^+ (t_{2g}) manifold into two sets of spin orbitals: a twofold degenerate set with Γ_7^+ symmetry and a fourfold degenerate set with Γ_8 symmetry. The Γ_{12}^+ (e_g) manifold is not split in the presence of SOC, but does acquire a new irreducible representation Γ_8^+ . Because orbitals from the original Γ_{25}^+ and Γ_{12}^+ manifolds share the same irreducible representations, they interact to form low- and high-energy pairs, akin to bonding and antibonding orbitals in a conventional MO diagram. The net result for the d^4 case is a completely filled Γ_8^+ manifold, which gives rise to the $J = 0$ state.

While the first scenario (direct metal-metal bonding) is well supported for the case of Li_2RuO_3 , it is not a likely explanation for our Ir^{5+} compounds, as NaIrO_3 and $\text{Sr}_3\text{CaIr}_2\text{O}_9$ show similar magnetic properties despite having wildly different Ir-Ir connectivity and internuclear distances. Further information can be obtained by examining the Ir–O bond lengths and octahedral distortions, as they provide direct signatures of orbital degeneracy. Fig. 6 compares the Ir–O bond lengths present in $\text{Sr}_3\text{CaIr}_2\text{O}_9$ and $\text{Sr}_3\text{CaRu}_2\text{O}_9$ using Gaussian kernel density estimates and illustrations to provide a visual understanding of the distortions present in these octahedra. Both compounds exhibit similar "three short, three long" distortions of the two octahedral metal sites, as discussed in section 2.1. These distortions are distinctly asymmetric, and thus inconsistent with a Jahn–Teller effect. The asymmetry is also likely not due to extended crystal field effects, as both Ir sites have similar extended coordination spheres (proximity to Ca^{2+} and Sr^{2+} ions). Rather, the observed asymmetry is likely a direct consequence of SOC, which, due to the mixing of bare-ion orbital identities, gives rise to asymmetric (i.e. direction-dependent) exchange. The fact that the octahedral distortions are amplified in $\text{Sr}_3\text{CaIr}_2\text{O}_9$ compared to $\text{Sr}_3\text{CaRu}_2\text{O}_9$ provides further support for the influence of SOC. Future spectroscopic experiments can directly probe for the excited states predicted by this model, and the optical signatures should also respond to application of a magnetic field, thus yielding a powerful method of investigating SOC–driven physics in these compounds.

4 Conclusions

Two new insulating honeycomb iridates, NaIrO_3 and $\text{Sr}_3\text{CaIr}_2\text{O}_9$, have been synthesized and characterized *via* diffraction experiments, magnetometry, and resistivity. In both compounds, iridium exists in a 5+ oxidation state, yielding a $5d^4$ electronic configuration. Magnetization measurements demonstrate that both compounds exhibit negligible magnetic susceptibility, implying that both of these compounds are in close proximity to either a $S = 0$ or a $J = 0$ magnetic state. Structural studies performed on $\text{Sr}_3\text{CaIr}_2\text{O}_9$ suggest SOC is the dominant energy scale in determining the ground magnetic state of Ir^{5+} compounds. Further spectroscopic experiments on these compounds will shed light on the nature of the magnetism observed in iridates.

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Compound ^{ref.}	μ_{eff} (μ_B/Ir)
$\text{Sr}_3\text{CaIr}_2\text{O}_9$ (this work)	0.18
Honeycomb NaIrO_3 (this work)	0.18
$\text{Sr}_5\text{La}_6\text{Ir}_4\text{O}_{24}$ ⁵	0.3
Post-perovskite NaIrO_3 ²²	0.28
$\text{Ba}_3\text{ZnIr}_2\text{O}_9$ ³⁰	0.2
Sr_2YIrO_6 ³¹	0.91

Table 1: Comparison of magnetic moments of NaIrO_3 and $\text{Sr}_3\text{CaIr}_2\text{O}_9$ with several previously reported Ir^{5+} compounds. With the exception of Sr_2YIrO_6 , all compounds exhibit small magnetic moments consistent with a $J = 0$ state in the bulk Ir^{5+} species, where the measured susceptibility is due to a small magnetic impurity or contribution from polycrystalline edge states.

Type	Compound ^{ref.}	Connectivity	Electronic configuration	Magnetic Properties
<i>Planar</i>	SrRu ₂ O ₆ ^{32,33}	edge-sharing	4d ³	AFM, $\mu_{eff} = 1.43(1) \mu_B$
	Ag ₃ LiRu ₂ O ₆ ³⁴	edge-sharing	4d ⁴	Weak, temperature-independent
	(Li/Na) ₂ RuO ₃ ³⁵	edge-sharing	4d ⁴	Weak, temperature-independent
	NaIrO ₃ (this work)	edge-sharing	5d ⁴	Weak, temperature-independent
	α -RuCl ₃ ³⁶	edge-sharing	4d ⁵	AFM, $\mu_{eff} = 2.2 \mu_B$
	Li ₂ RhO ₃ ³⁷	edge-sharing	4d ⁵	Spin-glassy AFM, $\mu_{eff} = 2.0 \mu_B$
	Li ₂ IrO ₃ ³⁸	edge-sharing	5d ⁵	AFM, $\mu_{eff} = 2.03 \mu_B$
	Na ₂ IrO ₃ ³	edge-sharing	5d ⁵	AFM, $\mu_{eff} = 2.02 \mu_B$
	Li ₂ PtO ₃ ³⁹	edge-sharing	5d ⁶	Unreported
	\mathcal{H} (1)-Li ₂ IrO ₃ ¹⁶	edge-sharing	5d ⁵	Anisotropic, $\mu_{eff} = 2.03 \mu_B$
<i>Harmonic</i>				
<i>Buckled</i>	Sr ₃ CaRu ₂ O ₉ ¹⁸	corner-sharing	4d ³	Unreported
	Sr ₃ CaIr ₂ O ₉ (this work)	corner-sharing	5d ⁴	Weak, temperature-independent
	<i>thin film</i> Ca _{0.5} Sr _{0.5} IrO ₃ ¹⁵	corner-sharing	5d ⁵	Unreported

Table 2: Comparison of known 4d and 5d honeycomb compounds organized by connectivity and primary metal ion. Basic magnetic properties are listed for reference.

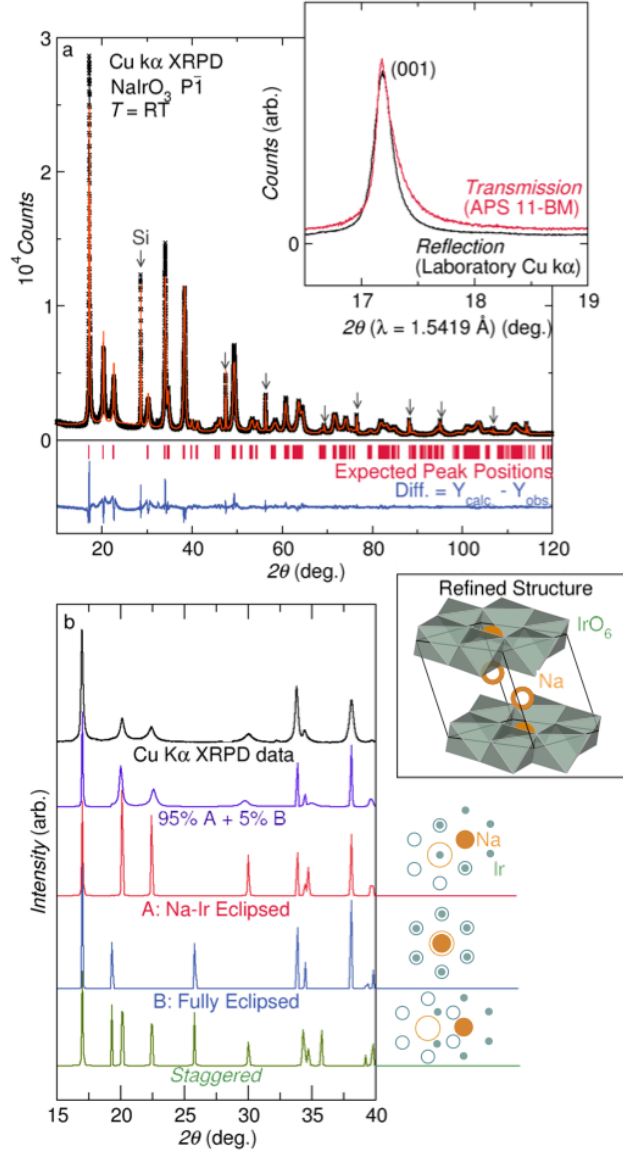


Figure 1: **a)** Laboratory XRPD data (black) collected on NaIrO_3 along with Rietveld refinement of the structure in $P\bar{1}$ to the data (red) and calculated difference between the model and the data (blue) are shown in the main frame. Stacking faults between honeycomb layers complicate the refinement, as can be seen from the missed intensities and peak shape mismatches between $2\theta = 15^\circ$ to 35° . In the inset, the observed profile of the (001) peak, as observed by laboratory XRPD data collected in reflection geometry, is compared to the profile observed from synchrotron XRPD data collected on APS 11-BM in transmission geometry. **b)** Laboratory XRPD data (black, top) along with simulations for three distinct stacking arrangements of adjacent NaIr_2 honeycomb layers (red, blue and green) are shown. A fourth simulation (purple) shows the pattern that results from having a 95% probability of an Na-Ir eclipsed stacking arrangement with a 5% probability of a fully eclipsed stacking fault, which provides a good description of the observed diffraction pattern. In the upper right-hand corner the structure of NaIrO_3 is shown, in which IrO_6 octahedra (green) share edges to form honeycomb layers with Na cations (orange) occupying the honeycomb "holes" and partially occupying the spaces sandwiched between Na and Ir sites in the honeycomb layers.

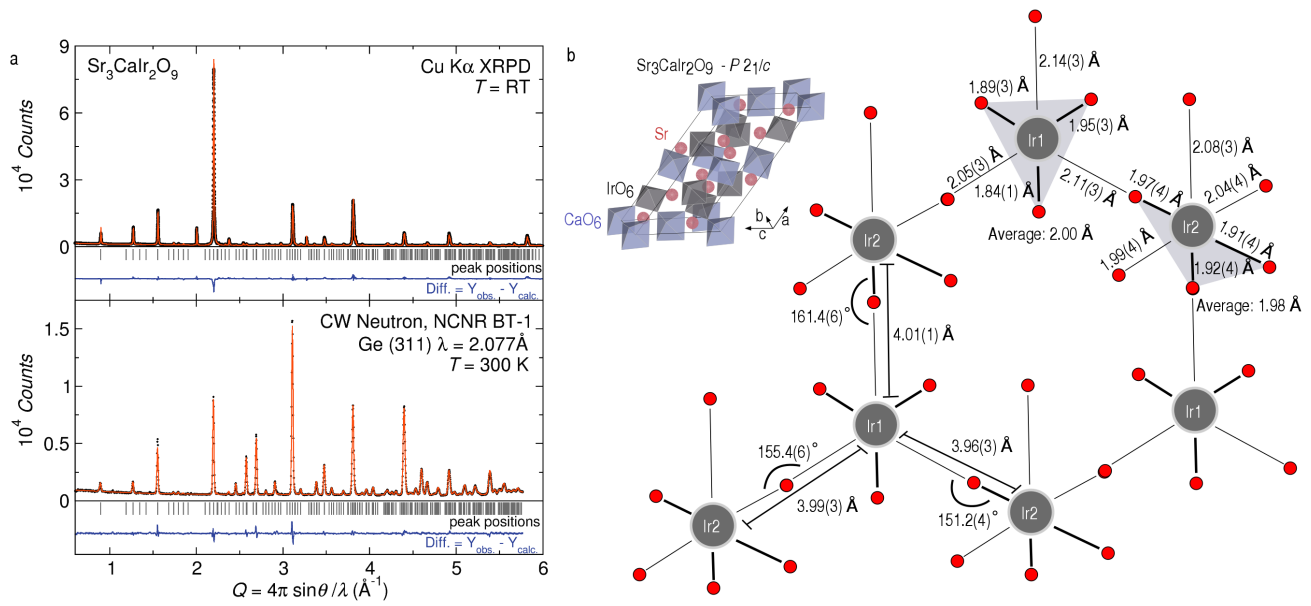


Figure 2: **a)** Rietveld refinement of the structure of $\text{Sr}_3\text{CaIr}_2\text{O}_9$ to laboratory XRPD data (top) and NCNR BT-1 NPD data (bottom) are shown. Data points are displayed in black, the fit in orange, and the difference between the data and fit, blue. Expected peak positions for the monoclinic cell are displayed as gray tick marks below the data in each pane. **b)** The unit cell of $\text{Sr}_3\text{CaIr}_2\text{O}_9$ is displayed along with a cutout of the honeycomb network formed by corner-sharing IrO_6 octahedra in the a - b plane. Ir-Ir distances and Ir-O-Ir bond angles are shown in the bottom left of the honeycomb structure, and Ir-O bond lengths for each of the two iridium sites are shown in the upper right hand corner. Despite the uniformity of the Ir-Ir distances in the honeycomb lattice, the IrO_6 octahedra exhibit significant asymmetric distortions, and these distortions are illustrated by gray triangles in the plane formed by the three most closely bonded oxygen atoms of each octahedron. These distortions are inconsistent with a Jahn-Teller effect, and instead are likely driven by strong SOC of the Ir^{5+} ion as described in the text.

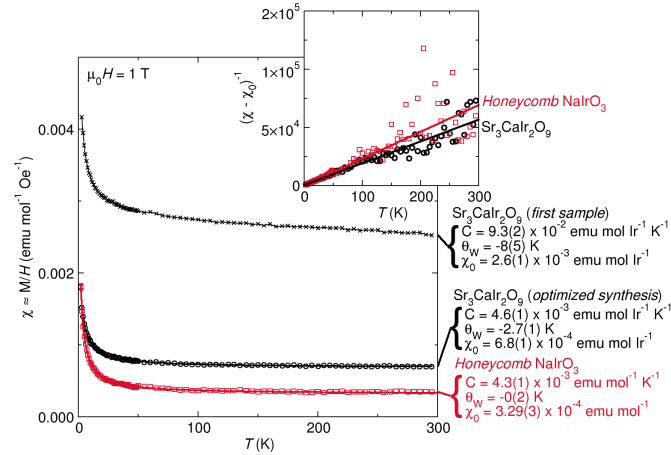


Figure 3: Magnetic susceptibility data collected on NaIrO_3 (red squares), the first sample of $\text{Sr}_3\text{CaIr}_2\text{O}_9$ (black crosses), and a sample of $\text{Sr}_3\text{CaIr}_2\text{O}_9$ produced via an optimized synthetic procedure (black circles) are shown. All magnetic data were collected on polycrystalline samples, and a correction for the temperature-independent susceptibility of the sample holder was applied to the data in order to extract real values of χ_0 for each sample. Two datasets are shown for $\text{Sr}_3\text{CaIr}_2\text{O}_9$ to illustrate that the observed sample χ_0 's are due to magnetic defects, the concentration of which can be substantially reduced through correct sample preparation. Linear fits to the χ_0 -corrected inverse susceptibility data for NaIrO_3 and $\text{Sr}_3\text{CaIr}_2\text{O}_9$, visible in the inset, were used to extract Curie constants C and Weiss temperatures Θ_W for NaIrO_3 and the optimized sample of $\text{Sr}_3\text{CaIr}_2\text{O}_9$. Both samples exhibit weak, temperature-independent susceptibility consistent with a small concentration of magnetic defects. These data demonstrate that the Ir^{5+} species in the bulk of both compounds have no unpaired electrons.

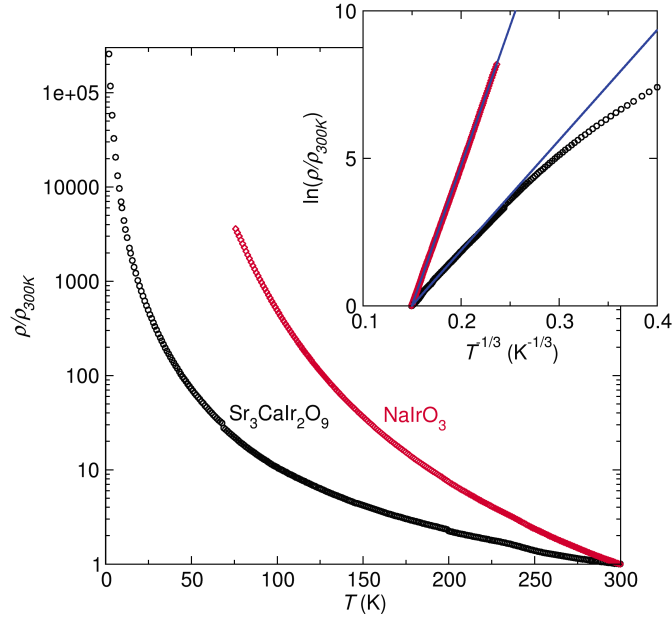


Figure 4: Four-probe resistivity data collected as a function of temperature on sintered polycrystalline bars of NaIrO_3 (red diamonds) and $\text{Sr}_3\text{CaIr}_2\text{O}_9$ (black circles) are shown. The increase in resistance of the pellets on cooling is a hallmark of electrically insulating materials. In the inset, the natural logarithm of the normalized resistance is plotted as a function of $T^{-1/3}$. Both datasets are linear in this plot in the high-temperature regime consistent with 2-dimensional variable-range hopping of impurity-localized charge carriers.

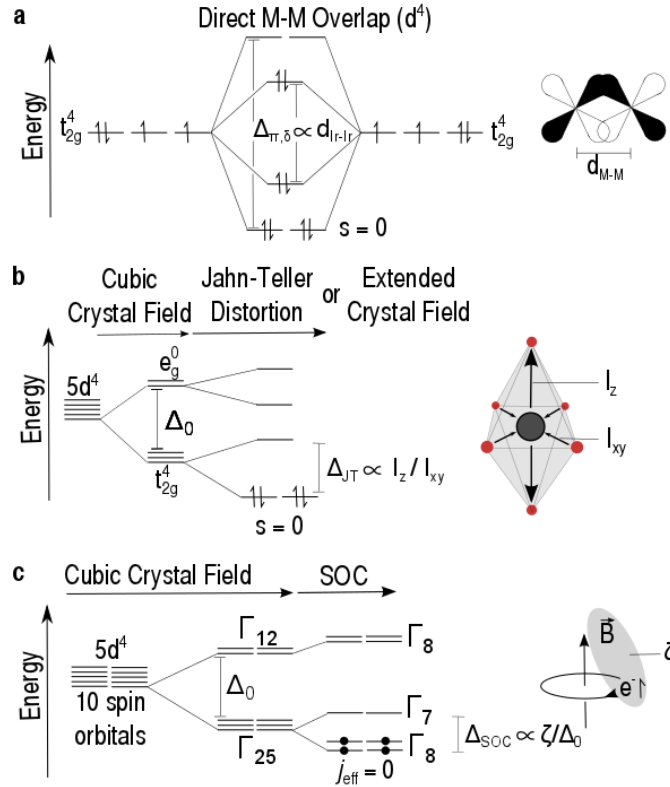


Figure 5: Potential origins of $S = 0$ and $J_{eff} = 0$ states in Ir^{5+} iridates. **a)** Direct π and δ overlap between adjacent iridium sites yields a nonmagnetic ($S = 0$) state by removing the threefold degeneracy of the isolated t_{2g} manifolds. The energy of the splitting, $\Delta_{\pi,\delta}$, is proportional to the distance between adjacent iridium sites, d_{Ir-Ir} . **b)** A local distortion of the cubic crystal field via Jahn-Teller distortion or interaction with cations outside the first coordination sphere (extended crystal field) splits the t_{2g} manifold by an energy proportional to the axial elongation (I_z/I_{xy}), which also yields a $S = 0$ state. **c)** The ground state of the d^4 configuration in a cubic crystal field can be split by SOC to yield a $J = 0$ ground state, which originates from $S = 1$ and $L = 1$ spin and orbital angular momenta. The upper bound to the magnitude of this splitting is proportional to the bare ion SOC constant ζ , relative to the crystal field Δ_0 .

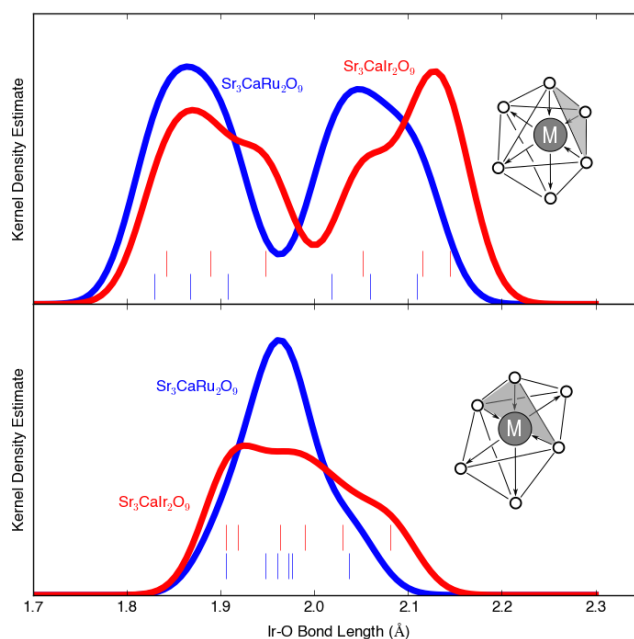


Figure 6: Visual comparison of M–O ($M = \text{Ir, Ru}$) bond lengths in $\text{Sr}_3\text{CaIr}_2\text{O}_9$ and $\text{Sr}_3\text{CaRu}_2\text{O}_9$. Gaussian kernel density estimates (KDE) computed on the Ir–O bond lengths present in $\text{Sr}_3\text{CaIr}_2\text{O}_9$ (red) and $\text{Sr}_3\text{CaRu}_2\text{O}_9$ (blue) provide a visual comparison of the octahedral distortions present in both compounds. The top and bottom panes contain data for the 1st and 2nd metal sites in each structure, respectively; in each pane, the solid lines represent KDE's and the small tick marks represent the actual M–O bond lengths. A distinct "3–short, 3–long" bond length pattern is observed in both compounds. Model octahedra in each pane show the shape of the distortion in both cases, with the plane formed by the closely bonded oxygen atoms highlighted in gray.